

## Effect Of A Denture Cleanser On Denture Base Resins – Original Article

### Abstract

**Purpose of the Study :** To evaluate and compare color stability and hardness of three different commercially available polymethylmethacrylate heat cure denture base resins following immersion in a denture cleansing agent, containing Sodium perborate.

**Materials and Method :** Total 45 disc-shaped specimens of three commercially available polymethyl methacrylate heat cure denture base resins, [15 specimens for each group (group A, T and D)] with 8 mm diameter and 3mm thickness were fabricated. The specimens were tested for baseline color, hardness and their optical density. 180 immersions (30 minutes of each immersion), simulating six months of denture cleansing were carried out in a sodium perborate based denture cleansing agent and above mentioned tests were repeated to study the effect of denture cleansing agent on the specimens.

**Results :** Paired sample t-test showed an increase in light transmission through the acrylic specimens ( $P < 0.001$ ) which demonstrated reduced optical density after immersion. Change in their color was also noted. Paired sample T-test in each group for hardness showed no significant difference before and after immersion ( $PX05;0.2$ ).

**Conclusion :** Long term immersion of denture base resin in sodium perborate based cleansing agent significantly affects its color.

### Key Words

Polymethylmethacrylate, denture cleanser, color stability, hardness

### Introduction:

Polymethyl-methacrylate (PMMA) constitutes usage in the widest spectrum of prosthodontic treatment options ranging from removable complete and partial dentures to implant retained prostheses. A prosthesis is subjected to a plethora of organisms in the oral cavity. The removal of biofilm deposited on a prosthesis can be accomplished with denture cleansers<sup>[1]</sup>. A wide variety of denture cleansing agents are used routinely and regularly by denture wearers. Hence, it becomes highly important to determine the effect of denture cleansers on the properties of acrylic resins. This study aimed to evaluate two important properties, colour stability and hardness of three different commercially available heat cure PMMA denture base resins caused by a denture cleansing agent containing sodium perborate.

### Materials And Method:

This study was carried out to compare the colour stability and hardness of commonly used, three different commercially available polymethylmethacrylate heat cure denture base resins (groups A, T and D).

### Fabrication Of Specimens

For the fabrication of specimens, dimensions of the customized metallic mold were selected in accordance with the size of diffuse reflectance spectra (DRS) assembly in the spectrophotometer (illustration 1). Thus, total 45 disc shaped wax-patterns of 8mm diameter and 3mm thickness were prepared using metallic mold. 15 wax-patterns for each of the three groups of specimens were invested in a mixture of dental stone and dental plaster. Each of the three flasks contained 15 disc shaped wax patterns (illustration 2). After wax elimination, resin was packed using compression molding technique in the molds created by these wax patterns following manufacturer's instructions. The resin was bench cured for 30 minutes, followed by heat



Illustration 1 : Customized Metallic Mold

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polymerization cycle in water bath at 730C for 90 minutes, which was followed by final curing 940C for 30 minutes. The flasks were allowed to bench cool for 24 hours before opening. The specimens were removed from the molds and immersed in distilled water at  $37 \pm 10$  C for 48 hours for residual monomer elimination. The excess resin was trimmed with a tungsten carbide bur using a handpiece at low speed (10,000 rpm), taking utmost care not to alter the dimensions. The specimens were checked for the dimensions using a digital vernier caliper (Insize, Amazon, India).

### Colour Analysis

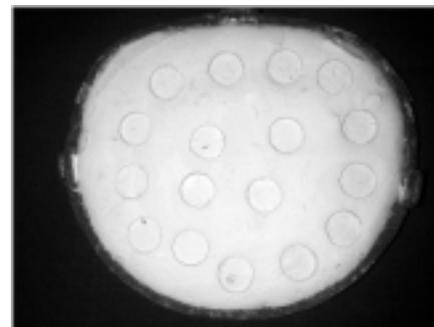


Illustration 2 : Molds Created By Wax-patterns

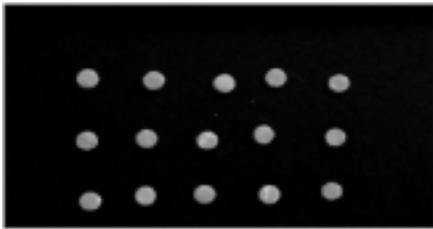


Illustration 3 : Specimens Arranged On Black Cardboard



Illustration 4 : Durometer Hardness Tester (Shore D)

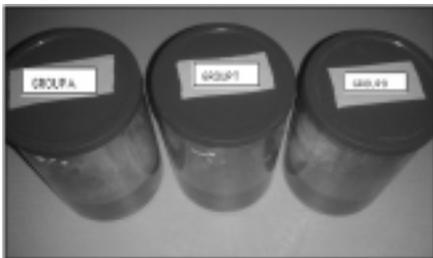


Illustration 5 : Immersion Of The Specimens In Denture Cleanser

The specimens of each of the three groups were numbered and analysis for colour and hardness was carried out to determine the baseline values. The spectrophotometer (Carrywin UV-Vis-NIR, Agilent Technologies, India) evaluated the amount of light transmitted through the specimens (optical density)<sup>[2]</sup>. For obtaining the CIE L\*a\*b\* values of the specimens, a digital technique was followed<sup>[3]</sup>. All the specimens were placed on a black cardboard surface (**illustration 3**) in order to acquire digital images. A white standard photograph paper was also used with the specimens as a calibration material to eliminate the environmental factors. A digital camera (DSLR- Nikon D 90) was fixed on a tripod, with a 40 cm object-camera distance (without flash). It was oriented perpendicular to the test samples to acquire the digital images. The images were taken at 11:00 AM, on a clear day, were saved in the tagged image

file format (TIFF), and were later resolved on a 24 bit resolution screen for further analysis using a commercial graphic software (Adobe Photoshop 6.0)<sup>[3],[4]</sup>. During the analysis, fixed circular areas with 74 pixels in diameter were selected from the middle third portion of each sample. The L\*, a\*, b\* values of these areas were measured three times by application of the histogram function of the software, and the mean values were recorded. These values were divided by the L\*, a\*, b\* values of the adjacent white photograph paper in order to eliminate the potential effects of the environmental factors, and the corrected L\*, a\*, b\* values of each specimen were recorded as the baseline colour measurement.

### Hardness Testing

Hardness testing was done using Durometer hardness tester (Shore D) according to ASTM-D2240 (American Standards of Testing and Materials) (**illustration 4**). The hardness value is determined by the Durometer indenter foot into the sample. If the indenter completely penetrates the sample, a reading zero is obtained, and if no penetration occurs, a reading of 100 results. The reading is dimensionless. Three measurements for indentation hardness were done on each specimen and their average values were recorded. This served as the hardness value before immersion in the denture cleansing solution.

### Immersion Procedure

Immersion of specimens from the three groups (T, A and D) were carried out at the same time in three different containers containing the cleansing solution (**illustration 5**). The effervescent cleanser was prepared according to the manufacturer's instructions, by adding one tablet to 200 ml of warm tap water and the immersion time being 30 minutes. After immersion for 30 minutes, the resin specimens were removed from the cleansing solutions, thoroughly washed in running water, dried with absorbent paper, and then the procedure of immersion was repeated. 180 immersions were performed over a period of 15 days, simulating 180 days of immersion, i.e.; 6 months of denture cleansing by the patient.

Following immersion process, colour and hardness analysis was carried out as mentioned previously. Colour

differences ( $\Delta E$ ) between the measurements (before and after immersion in cleanser solution), in terms of L\*, a\* and b\*, were calculated using following equation:<sup>[5]</sup>

in which  $\Delta L$ ,  $\Delta a$  and  $\Delta b$  are the differences of L, a and b values before and after immersion.

The levels of color change ( $\Delta E$ ) have been quantified by the National Bureau of Standards (NBS) with the NBS units of color difference. NBS units are expressed by the following formula: NBS unit =  $\Delta E \times 0.92$

Critical Marks Of Color Difference According To National Bureau Of Standard (1955)

Critical Marks Of Color Difference	Textile Terms (Nbs Unit)
Trace	0.0-0.5
Slight	0.5-1.5
Noticeable	1.5-3.0
Appreciable	3.0-6.0
Much	6.0-12.0
Very Much	>12.0

### Statistical Analysis

Data was summarized as Mean  $\pm$  SD. Paired sample t-test was applied to groups [T, A and D] and comparison was done using One way Analysis of Variance (ANOVA), following which post hoc Tukey's test was used. Paired sample t-test reveal significant change in %T before and after immersion in each group ( $P < 0.001$ ), which showed a reduction in optical density of the specimens after immersion in the denture cleanser (**Table 1**). Multiple comparisons, Tukey HSD test revealed that amongst the three groups, the color difference was not significant ( $P > 0.3$ ) (**Table 2**). Paired sample T-test in each group for hardness showed that there was no significant difference before and after immersion ( $P \geq 0.2$ ) (**Table 3**).

### Results:

The results depicted that the change in optical density was least in group T, followed by group A and D respectively. Moreover, the CIE L\*a\*b\* values of the specimens that were obtained with the help of digital camera using a commercial graphic software also revealed the same color change. The change in hardness was not significant.

### Discussion:

Mechanical denture cleansing methods are commonly aided by chemical denture cleansing agents which contain sodium

Table 1: Paired Sample T-test Reveal Significant Change In % Before And After Immersion

Paired Sample T-test							
Group = A							
		Mean	N	Std. Deviation	Std. Error Mean	Mean Difference	P Value
	% Transmission Visible Range Pre	18.89	15	9.143	2.361	9.29	<0.0001
	% Transmission Visible Range Post	28.18	15	9.807	2.532		
	% Transmission Ultraviolet Region Pre	13.79	15	7.161	1.849	5.57	<0.0001
	% Transmission Ultraviolet Region Post	19.36	15	7.693	1.986		
Group = T							
		Mean	N	Std. Deviation	Std. Error Mean	Mean Difference	P Value
	% Transmission Visible Range Pre	14.13	15	3.480	0.898	5.81	<0.0001
	% Transmission Visible Range Post	19.94	15	4.425	1.143		
	% Transmission Ultraviolet Region Pre	11.70	15	3.309	0.854	3.00	<0.0001
	% Transmission Ultraviolet Region Post	14.70	15	3.724	0.962		
Group = D							
		Mean	N	Std. Deviation	Std. Error Mean	Mean Difference	P Value
	% Transmission Visible Range Pre	17.74	15	6.957	1.796	12.72	<0.0001
	% Transmission Visible Range Post	30.45	15	9.862	2.546		
	% Transmission Ultraviolet Region Pre	15.15	15	5.738	1.482	10.74	<0.0001
	% Transmission Ultraviolet Region Post	25.89	15	8.485	2.191		

Table 2 : Tukey Hsd Test Reveals That Amongst The Three Groups, The Colour Difference Is Not Significant (P>0.3).

Post Hoc Tests							
Multiple Comparisons							
Tukey Hsd							
Dependent Variable	(I) Group	(J) Group	Mean Difference (I-J)	Std. Error	P Value	95% Confidence Interval	
						Lower Bound	Upper Bound
L*	T	A	-1.40	1.987	0.762	-6.23	3.43
	T	D	-2.67	1.987	0.380	-7.49	2.16
	A	D	-1.27	1.987	0.800	-6.09	3.56
A*	T	A	0.80	1.783	0.895	-3.53	5.13
	T	D	2.53	1.783	0.339	-1.80	6.86
	A	D	1.73	1.783	0.598	-2.60	6.06
B*	T	A	-0.53	1.493	0.932	-4.16	3.09
	T	D	-1.27	1.493	0.675	-4.89	2.36
	A	D	-0.73	1.493	0.876	-4.36	2.89
E	T	A	-0.18	2.212	0.996	-5.56	5.19
	T	D	-0.66	2.212	0.952	-6.04	4.71
	A	D	-0.48	2.212	0.974	-5.85	4.89

Table 3 : Paired Sample T-test In Each Group For Hardness Shows That There Is No Significant Difference Before And After Immersion (P = 0 . 2 )

Paired Sample T-test							
		Mean	N	Std. Deviation	Std. Error Mean	Mean Difference	P Value
Group = A	Durometer Hardness Before	90.67	15	1.23	0.319	-0.200	0.271
	Durometer Hardness After	90.47	15	1.25	0.322		
Group = D	Durometer Hardness Before	90.33	15	1.05	0.270	-0.133	0.737
	Durometer Hardness After	90.20	15	1.37	0.355		
Group = T	Durometer Hardness Before	89.47	15	0.99	0.256	-0.133	0.164
	Durometer Hardness After	89.33	15	0.90	0.232		

perborate and alkaline peroxide in tablet form. These effervescent tablets are classified as chemical soak type products. When dissolved in water, sodium perborate decomposes to form an alkaline peroxide solution. This peroxide solution subsequently releases oxygen,

enabling a mechanical cleaning by the oxygen bubbles in addition to the chemical cleaning. Efficacy of these agents in removing stains and reducing formation of biofilm on the irregularities of denture surfaces have been reported.<sup>[6]</sup> Evaluating the effect of denture cleansers

on the properties of the resin is highly important. Denture base polymers are susceptible to color shifting if the cleansing solutions are not used correctly. The whitening effect relates to the high temperature of the water used in the solution<sup>[7]</sup>. The effervescent tablets are efficient in removing biofilm and stains, but the alkaline peroxide solution can alter the resin properties if used inappropriately<sup>[7]</sup>.

Colour changes were evaluated using the CIE L\*a\*b\* colorimetric system,<sup>[8],[9]</sup> a uniform 3-dimensional system that has been widely used for the determination of chromatic differences by translating combinations of differences into mathematical data. NBS units were also used in this study, because the NBS parameter is important for color comparison and quality control functions. For the acrylic resins evaluated in this study, L\* values decreased for all the groups. The same change was noted with the help of spectrophotometer, which showed increase in light transmission (%T) through the specimens after immersion in the denture cleanser, due to the reduced optical density of the specimens. The spectrophotometer is an instrument that separates electromagnetic radiation into its component wavelengths and selectively measures the intensity of radiation after passing through a sample. The absorption or transmission of specific wavelengths is characteristic for a substance,<sup>[10]</sup> and a spectral analysis serves as a “fingerprint” of the compound.

Colour change is an indicator of aging or damage to dental materials,<sup>[11],[12]</sup> it increases as immersion time increases. Sarac et al and Purnaveja et al reported that denture cleanser can cause whitening or bleaching, loss of soluble components, or increase in water absorption of PMMA resin materials,<sup>[13],[14]</sup>. The PMMA resins evaluated in this study tend to change colour during long periods of immersion in distilled water. This could be a result of the leaching out of coloring material. Previous reports have shown that color change in denture base materials is caused by changes in the matrix of the material and the staining effect of external colorants. Solubility, water sorption, leakage, surface roughness, and chemical degradation of denture base material may also cause change in colour. This could explain why L\*, a\*, and b\* values increased or decreased.

Hardness is a term used to describe the resistance of a material to plastic deformation typically measured under indentation load<sup>[15]</sup>. Concerning the results of the present study, a statistical analysis showed a non-significant difference before and after immersion in the denture cleanser (for each group). The absence of any effect of immersion solutions on the surface hardness of acrylic resins could be due to the presence of cross-linking material which reduces the denture base solubility to organic solvents,<sup>[16],[17]</sup>.

Additional investigations regarding the influence of denture cleanser on the other properties like surface roughness, flexural strength of these materials are indicated. Furthermore, additional studies on the relationship between composition of denture base acrylic resins and their colour stability and hardness are necessary to understand the effects of aging of denture cleansers on the mechanism of colour change and change in hardness. Studies using different concentration of cleansing agents also need to be done.

#### Conclusion:

Within the limitations of the present study, it can be concluded that sodium perborate based denture cleansing agent causes a change in the color of the PMMA denture base material, when immersion is done over a long period of time, but does not cause a significant change in hardness. So this kind of denture cleansing agents should be used less frequently, and according to manufacturer's instructions.

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